



## BAB V

### KESIMPULAN DAN SARAN

#### 5.1 Kesimpulan

1. Suhu dan lama waktu aktivasi yang digunakan pada penelitian ini belum mampu memacu perkembangan struktur pori pada karbon aktif.
2. Hasil SEM menunjukkan bahwa variasi aktivasi yang digunakan belum memacu secara utuh perkembangan pori pada permukaan karbon aktif.
3. Hasil Analisa BET menunjukkan perkembangan luas permukaan yang kecil, yaitu sekitar  $6\text{-}30 \text{ m}^2/\text{g}$ . Berdasarkan adsorpsi isoterm dan *pore size distribution*, dapat diketahui bahwa karbon aktif yang dihasilkan memiliki bentuk pori relatif mesopori.
4. Hasil FTIR menunjukkan bahwa adanya perubahan gugus fungsi dari kulit salak, *hydrochar*, hingga menjadi karbon aktif. Selain itu, terlihat adanya pengurangan peak pada spektrum gugus fungsi dari *hydrochar* ketika diaktivasi menjadi karbon aktif. Bila dibandingkan dengan karbon aktif komersial, Hasil FTIR dari karbon aktif yang diperoleh memiliki spektrum yang mulai mendekati dengan karbon aktif komersial.

#### 5.2 Saran

1. Adanya pengecekan alat pada reaktor superkritik yang digunakan agar tidak terdapat penyumbatan ataupun *leak* yang dapat mengganggu hasil sampel.
2. Penggunaan suhu operasi yang cukup tinggi dan waktu aktivasi yang cukup lama agar dapat memacu perkembangan pori pada sampel karbon aktif.
3. Adanya perlakuan pada sampel *hydrochar* dengan menggunakan asam sitrat atau *cerous chloride* sebagai katalis sebelum diaktivasi untuk memacu perkembangan distribusi ukuran pori pada sampel karbon aktif.

## DAFTAR PUSTAKA



*Biomassa Sebagai Sumber Energi Terbarukan.* (2008, 01 14). Retrieved 7 25, 2018, from DW AKADEMIE: <http://www.dw.com/id/biomassa-sebagai-sumber-energi-terbarukan/>

A. AL Othman, Z. (2012). A Review: Fundamental Aspects of Silicate Mesoporous Material. *Materials*, 2874-2902.

Andreas, A., Putranto, A., & Sabatini, T. C. (2015). Sintesis Karbon Aktif dari Kulit Salak dengan Aktivasi  $K_2CO_3$  sebagai Adsorben Larutan Zat Warna Metilen Biru .

Anovitz, L. M., & Cole, D. R. (2015). Characterization and Analysis of Porosity and Pore Structure. *Mineralogical Society of America*, 61-164.

Arie, A. A., Vincent, & Putranto, A. (2016). Activated carbons from KOH-activation of salacca peels as low cost potential adsorbents for dye removal . *Advanced Materials Letter*, 226-229 .

Bandosz, T. J. (2006). Activated Carbon Surfaces in Environmental Remediation. New York: Elsevier Science & Technology Books.

Bandosz, T. J. (2006). Activated Carbon Surfaces in Environmental Remediation. New York: ELSEVIER.

Bansal, R. C., & Goyal, M. (2005). Activated Carbon Adsorption. Boca Raton: Taylor & Francis Group.

Budinova, T., Ekinci, E., Yardim, F., Grimm, A., Bjornbom, E., Minvkova , V., et al. (2006). Characterization and application of activated carbon produced by  $H_3PO_4$  and water vapor activation. *Fuel Processing Technology*, 899-905.

Castañeda, D., Corredor, L., Baquero, M. C., & Rodríguez, L. I. (2004). Activated Carbon from Oil Palm Shell Obtained by Supercritical Water Treatment.

Chen, H. (2014). Biotechnology of Lignocellulose. Beijing: Chemical Industry Press.

Coronella, C. J., Lynain, J. G., Reza, M. T., & Uddin, M. H. (2013). Hydrothermal Carbonization of Lignocellulosic Biomass.

Cuhadaroglu, D., & Uygun, O. A. (2008). Production and characterization of activated carbon. *African Journal of Biotechnology*, 3703-3710.

Demiral, I., & Samdan, C. A. (2016). Preparation and Characterisation of Activated Carbon from Pumpkin Seed Shell using  $H_3PO_4$ . *Anadolu University Journal of Science and Technology A- Applied Sciences and Engineering*, 125-138.

Domac, J., & Trossero, M. (2008). Industrial Charcoal Production, 9.

Ebbesen, T. W., & Takada, T. (1995). Carbon volume 33.

- Fang, Q.-R., Makal, T. A., Young, M. D., & Zhou, H.-C. (2010). Recent Advances in The Study of Mesoporous Metal-Organic Frameworks. *Comments on Inorganic Chemistry*, 165-195.
- Fengel, D., & Wegener, G. (1995). Kayu: Kimia, Ultra Struktur dan Reaksi-Reaksi.
- Fengel, D., & Wegener, G. (1995). Kayu: Kimia, Ultra Struktur dan Reaksi-Reaksi. 669-729.
- Ferdi Schuth, K. S. (2002). Handbook of Porous Solid. Weinheim: Federal Republic of Germany.
- Gomez-Serrano, V. d. (2005). Preparation of activated carbons from chesnut wood by phosphoric acid chemical activation.
- gunam, I. B., Buda, K., & guna, I. M. (2010). Pengaruh Perlakuan Delignifikasi dengan Larutan NaOH dan Konsentrasi Substrat Jerami Padi Terhadap Produksi Enzim Selulase dari Aspergillus Niger NRRL A-264. *BILOGI XIV*, 55-61.
- Harry Marsh, F. R.-R. (2006). Activated Carbon. Elsevier Science & Technology Books.
- Hendri, Z., Arianingrum, R., & Zuhdi, B. M. (2010). Penerapan Teknologi Pemanfaatan Kulit Salak pada Produk Keramik Guna Peningkatan Usaha Kerajinan Keramik. *Inotek*.
- Instruments, Q. (2013, 06 28). *Calculation of BET Area of Microporous Material with Automated Software*. Retrieved 07 26, 2018, from Azom: <https://www.azom.com/article.aspx?ArticleID=9434>
- Ioannidou, O., & Zabanitou, A. (2006). Agricultural residues as precursors for activated carbon production. *Renewable & Sustainable Energy Reviews*.
- Jain, A., Balasubramanian, R., & Srinivasan, M. P. (2015). Hydrothermal Conversion of Biomass Waste to Activated Carbon with High Porosity. *Chemical Engineering Journal*.
- Jankowska, H., Swiatkowski, A., & Choma, J. (1991). Active Carbon. Warsaw: Ellis Hardwood.
- Kalderis, D., Kotti, M. S., Méndez, A., & Gascó, G. (2014). Characterization of hydrochars produced by hydrothermal carbonization of rice husk. *solid earth*, 477-483.
- Kang, S., Li, X., Fan, J., & Chang, J. (2012). Characterization of Hydrochars Produced by Hydrothermal Carbonization of Lignin, Cellulose, D-Xylose, and Wood Meal. *American Chemical Society*, 9023-9031.
- Kobya, M., Demirbas, E., Senturk, E., & Ince, M. (2005). Bioresource Technol. Adsorption of heavymetal ions from aqueous solutions by activated carbon prepared from apricot stone, 1518-1521.
- Kopac, T., & Toprak, A. (2007). Preparation of activated carbons from zonguldak region coals by physical and chemical activations for hydrogen sorption. *Hydrogen Energy*.

- Krumeich, F. (2017). *electron microscopy*. Retrieved from electron microscopy: <http://www.microscopy.ethz.ch/sem.htm>
- Kumar, P., Barrett, D. M., Delwiche, M. J., & Strocvc, P. (2009). Method for Pretreatment of Lignocellulosic Biomass for Efficient Hydrolysis and Biofuel Production. *I&EC*.
- Kurniawan, F., Wongso, M., Ayucitra, A., Soetaredjo, F. E., Angkawijaya, A. E., Ju, Y.-H., et al. (2015). Carbon microspher from water hyacinth for supercapacitor electrodea. *Taiwan Institute of Chemical Engineers*, 197-201.
- Lestari, R. (2013). Fruit Quality Changes of Salak "Pondoh" Fruits (*Salacca zalacca* (Gaertn.) Voss) during Maturation and Ripening. *Journal of Food Research*.
- Lin, Y., Ma, X., Peng, X., & Yu, Z. (2016). A Mechanism Study on Hydrothermal Carbonizatioon of Waste Textile. *energy&fuels*.
- Marsh, H., & Rodriguez-Reinoso, F. (2006). Activated Carbon. Elsevier Science & Technology Books.
- Material Evaluation and Engineering, I. (2014). *Material Evaluation and Engineering, Inc*. Retrieved from Material Evaluation and Engineering, Inc: <http://www.mee-inc.com/hamm/scanning-electron-microscopy-sem/>
- Mattson, J. S., & Mark, H. B. (1971). *Activated Carbon : Surface Chemistry and Adsorption from Solution*. M. Dekker.
- Moller, M., Nilges, P., Harnisch, F., & Schroder, U. (2011). Subcritical Water as Reaction Environment: Fundamentals of Hydrothermal Biomass Transformation. *CHEMSUSCHEM*.
- Mufioz-Guillena, M., Illan-Gomez, M., Martin-Martinez, J., Linares-Solano, A., & Salinas-Martinez de Lecea, C. (1992). *Energy Fuels volume 6*.
- Nicolet, T. (2001). Fourier Transform Infrared Spectrometry. Retrieved from Thermo Nicolet Corporation.
- Othmer, K. (1979). Encyclopedia of Chemical Technology 3rd Edition. John Wiley & Sons.
- Othmer, K. (1992). Encyclopedia of Chemical Technology:Bearing. 4.
- Pierson, H. O. (1993). Handbook of Carbon, Graphite, Diamond and Fullerenes. Albuburque: Noyes Publications.
- Pierson, H. O. (1993). Handbook of Carbon, Graphite, Diamond and Fullerenes: Properties, Processing and Applications. Albuquerque, New Mexico: NOYES PUBLICATIONS.
- Poulopoulos, S., & Inglezakis, V. (2006). Adsorption, Ion Exchange and Catalysis: Design of Operations and Environmental Applications. Elsevier.
- Qiong Cai, Z.-H. H.-B. (2004). Preparation of Activated Carbon Microspheres from Phenolic-Resin by Supercritical Water Activation. *Carbon* 42, 775-783.

- Roop Chand Bansal, M. G. (2005). Activated Carbon Adsorption. CRC Press Taylor & Francis Group.
- Rosalina, Tedja , T., Riani, E., & Sugiarti, S. (2016). Pengaruh Aktivasi Fisika dan Kimia Arang Aktif Buah Bintaro Terhadap Daya Serap Logam Berat Krom.
- Rouquerol, J., & Llewellyn, P. L. (2007). Is the BET Equation Applicable to Microporous Adsorbents ? *Studies in Surface Science and Catalyst*, 49-56.
- S., O. Z., S., P. P., & Faizal, M. (2012). Pengaruh Konsentrasi Asam dan Waktu pada Proses Hidrolisis dan Fermentasi Pembuatan Bioetanol dari Alang-Alang. *Jurnal Teknik Kimia No. 2, Vol. 18*, 52-62.
- S.Kang, X.Li, J.Fan, & Chang, J. (2012). Characterization of Hydrochars Produced by Hydrothermal Carbonization of Lignin, Cellulose, D-xylose, and Wood Meal. *Industrial & Engineering Chemistry Research* 51, 9023-9031.
- Salvador Palacios, F. (1998). Patent No. EP 0974 553 A1. Madrid.
- Salvador, F., Jesu'sSa'nchez-Montero, M., Montero, J., & Izquierdo, C. (2008). Activated Carbon Fibers Prepared from a Phenolic Fiber by Supercritical Water and Steam Activation. *Physic Chemical C*, 20057-20064.
- Savova, D., Apak, E., Ekinci, E., Yardim, F., Petrova, N., & Budinova, T. (2001). Biomass conversion to carbon adsorbents and gas. *Biomass Bioenergy*.
- Sevilla, M., & Fuertes, A. (2009). The Production of Carbon Material by Hidrotermal Carbonization of Cellulose. *Elsevier*, 2281-2289.
- Smisek, M., & Cerny, S. (1970). Active Carbon Manufacture, *Properties and Applications*. Elsevier.
- Srinivasakannan, C., & Zailani Abu Bakar, M. (2004). Production of activated carbon from rubber wood sawdust. *Biomass Energy*, 89-96.
- Susanti, R. F., Kevin, G., Erico, M., Kevien, Andreas, A., Kristianto, H., et al. (2018). Delignification, Carbonization Temperature and Carbonization Time Effects on the Hydrothermal Conversion of Salacca peel. *American Scientific Publisher*, 7263-7268.
- Tsai, W. d. (2001 ). Adsorption of acid dye onto activated carbons prepared from agricultural waste bagasse by ZnCl<sub>2</sub> activation.
- Vinita, R., & Handoko, T. (2017). Pemisahan Selulosa dan Lignin Buah Bintaro dengan Metode Pretreatment Alkali. Bandung: Departement of Chemical Engineering , Parahyangan Catholic Univeristy.
- Ying, L., Zheng-hong, H., Fei-yu, K., & Bao-hua, L. ( 2010). Preparation of ACMs from phenolic resin with metal compunds by sub- and supercritical water activation. *New Carbon Materials*, 109–113.

- Z.Fang, T.Sato, Jr, R. S., Inomata, H., Arai, K., & Kozinski, J. (2008). Reaction Chemistry and Phase Behavior of Lignin in High Temperature and Supercritical Water. *Bioresource Technology* 99, 3424-3430.
- Zhang, L., Xu, C., & Champagne, P. (2009). Recent advances in thermo-chemical conversion of biomass. *Energy Conversion and Management*.
- Zhao, Y., Li, W., Zhao, X., Wang, D. P., & Liu, S. X. (2013). Carbon spheres obtained via citric acid catalysed. *Materials Research Innovations* , 546-551.
- Zheng, Y., Zhao, J., Xu, F., & Li, Y. (2014). Pretreatment of Lignocellulosic biomass for enhanced biogas production. *ELSEVIER*, 1-19.
- Zhong, Zhuo-Ya, Yang, Q., Li, X.-M., Luo, K., Liu, Y., et al. (2012). Preparation of Peanut Hull-Based Activated Carbon By Microwave-Induced Phosphoric Acid Activator and Its Application In Remozal Brilliant Blue Adsorption. *Elsevier*, 178-183.