



BAB V

KESIMPULAN DAN SARAN

5.1 Kesimpulan

1. Adsorben yang dihasilkan terdiri dari CNT, CNS, hematite, dan magnetite
2. Komposit CNT yang dihasilkan memiliki diameter sekitar 0,290 – 0,318 μm
3. Kondisi pH optimum diperoleh pada pH 6 dengan % removal sebesar 55,658%
4. % removal dari adsorben komposit CNT yang dihasilkan lebih baik dibandingkan dengan karbon aktif
5. Semakin tinggi konsentrasi awal ion logam berat Ni(II), semakin kecil % removal yang dihasilkan
6. Semakin besar massa adsorben yang digunakan, semakin besar % removal yang dihasilkan
7. Semakin tinggi temperatur operasi, semakin kecil % removal yang dihasilkan
8. Model kinetika adsorpsi yang paling sesuai mengikuti model kinetika adsorpsi pseudo orde dua
9. Model isothermal adsorpsi yang paling sesuai adalah model Langmuir yang menghasilkan kapasitas adsorpsi (q_m) sebesar 50,25 mg logam Ni²⁺/g komposit CNT dan konstanta Langmuir sebesar 0,059 L/mg logam Ni²⁺
10. Komposit CNT yang dihasilkan mempunyai kemampuan adsorpsi yang baik

5.2 Saran

1. Bentuk besi selongsong untuk mengeluarkan material karbon nano dari dinding reaktor perlu dipertimbangkan agar mempermudah proses pengeluaran sampel.
2. Perlu dipikirkan cara untuk memperbesar %*yield* dari komposit CNT yang dihasilkan
3. Perlu menemukan metode analisa kadar logam Ni(II) yang lebih cepat

DAFTAR PUSTAKA



- [1] "<http://ilmupengetahuanumum.com/10-negara-penghasil-nikel-terbesar-di-dunia/>," [Online].
- [2] D. Lakherwal, "Adsorption of Heavy Metals: A Review," *International Journal of Environmental Research and Development*, 2014.
- [3] "<http://www.kelair.bppt.go.id/>," [Online].
- [4] S. Deng, "Sorbent Technology," *Encyclopedia of Chemical Processing*, 2006.
- [5] M. I. Kandah and J. L. Meunier, "Removal of nickel ions from water by multi-walled carbon nanotubes," *Journal of Hazardous Materials*, 2006.
- [6] X. Liu, M. Wang, S. Zhang and B. Pan, "Application potential of carbon nanotubes in water treatment: A review," *Journal of Environmental Sciences*, 2013.
- [7] anonim, "nanocs product," 2013. [Online]. Available: www.nanocs.com. [Accessed 18 May 2017].
- [8] B. Yulianto, 2005. [Online]. Available: <http://www.nano.lipi.go.id/utama.cgi?cetakartikel&1073086044>.
- [9] A. Andreas, H. Kristianto, W. Willianti and N. Orlando, "SINTESIS MATERIAL KARBON NANO DARI MINYAK GORENG KELAPA SAWIT DENGAN METODE NEBULIZED SPRAY PYROLYSIS," 2016.
- [10] C. Chen and X. Wang, "Adsorption of Ni(II) from Aqueous Solution Using Oxidized Multiwall Carbon Nanotubes," *Ind. Eng. Chem. Res.*, 2006.
- [11] H. Hasar, "Adsorption of nickel(II) from aqueous solution onto activated carbon prepared from almond husk," *Elsevier*, 2002.
- [12] E. Worch, *Adsorption Technology In Water Treatment*, Berlin: De Gruyter, 2012.
- [13] A. Wiyarsi and E. Priyambodo, "Pengaruh Konsentrasi Kitosan dari Cangkang Udang Terhadap Efisiensi Penjerapan Logam Berat," Yogyakarta.
- [14] E. W. Hajar, R. S. Sitorus, N. Mulianingtiyas and F. J. Welan, "Efektivitas Adsorpsi Logam Pb²⁺ dan Cd²⁺ Menggunakan Media Adsorben Cangkang Telur Ayam," Konversi, Samarinda, 2016.
- [15] T. S. Y. Choong, T. N. Wong, T. G. Chuah and A. Idris, "Film-pore-concentration-dependent surface diffusion model for the adsorption of dye onto palm kernel shell activated carbon," *Journal of Colloid and Interface Science*, 2006.

- [16] R. D. Noble and P. A. Terry, *Principles of Chemical Separations with Environmental Applications*, Cambridge: Cambridge University Press, 2004.
- [17] K. V. Kumar, K. Subanandam, V. Ramamurthi and S. Sivanesan. [Online].
- [18] E. F. Mohamed, "Removal of Organic Compounds from Water by Adsorption and Photocatalytic Oxidation," Institute National Polytechnique de Toulouse, 2011.
- [19] D. M. Ruthven, *Principles of Adsorption and Adsorption Processes*, New York: John Wiley & Sons, Inc., 1984.
- [20] I. A. Oke, N. O. Olarinoye and S. R. A. Adewusi, "Adsorption kinetics for arsenic removal from aqueous solutions by untreated powdered eggshell," *Springer Science*, 2007.
- [21] Q. S. Liu, T. Zheng, P. Wang, J. P. Jiang and N. Li, "Adsorption isotherm, kinetic and mechanism studies of some substituted phenols on activated carbon fibers," *Chemical Engineering Journal Elsevier*, 2009.
- [22] T. V. Thuan, B. T. P. Quynh, T. D. Nguyen, V. T. T. Ho and L. G. Bach, "Response surface methodology approach for optimization of Cu²⁺, Ni²⁺ and Pb²⁺ adsorption using KOH-activated carbon from banana peel," *Elsevier*, 2016.
- [23] L. Dong, W. Liu, R. Jiang and Z. Wang, "Study on the adsorption mechanism of activated carbon removing low concentrations of heavy metals," *Desalination and Water Treatment*, 2015.
- [24] A. Sharma, D. Anghore, R. Awasthi, S. Kosey, S. Jindal, N. Gupta, D. Raj and R. Sood, "REVIEW ON CURRENT CARBON NANOMATERIALS AND OTHER NANOPARTICLES TECHNOLOGY AND THEIR APPLICATIONS IN BIOMEDICINE," *WORLD JOURNAL OF PHARMACY AND PHARMACEUTICAL SCIENCES*, vol. 4, no. 12, 2015.
- [25] M. S. Dresselhaus, G. Dresselhaus and P. Avouris, *Carbon Nanotubes Synthesis, Structure, Properties, and Applications*, Berlin: Springer, 1996.
- [26] A. Kruger, *Carbon Materials and Nanotechnology*, Germany: WILEY-VCH, 2010.
- [27] B. J. H. Suslick and S. Kenneth, "Applications of Ultrasound to the Synthesis of Nanostructured Materials," 2010.
- [28] E. T. Thostenson, Z. Ren and T.-W. Chou, "Advances in the science and technology of carbon nanotubes and their composites: a review," 2001.
- [29] F. Asip, R. Mardhiah and H. , "UJI EFEKTIFITAS CANGKANG TELUR DALAM MENGADSORBSI ION Fe DENGAN PROSES BATCH," *Jurnal Teknik Kimia*, vol. 15, 2008.

- [30] M. Satake, "Spectrophotometric determination of nickel by adsorption of nickel dimethylglyoximate on naphthalene," 1979.
- [31] "Sigma Aldrich," 2016. [Online]. Available: <http://www.sigmaaldrich.com/catalog/product/sial/162574?lang=en®ion=ID>.
- [32] D. A. Skoog, F. J. Holler, S. R. Crouch and D. M. West, Principles of Instrumental Analysis 9th Edition, Brooks/Cole, Cengage Learning, 2004.
- [33] M. Maleque, M. R. Hasan, F. Hossen and S. Safi, "Development and validation of a simple UV spectrophotometric method for the determination of levofloxacin both in bulk and marketed dosage formulations," *Journal of Pharmaceutical Analysis*, 2012.
- [34] A. Nieto-Marquez, R. Romero and A. Romero, "Carbon nanospheres: synthesis, physicochemical properties and applications," 2010.
- [35] S. E.-S. Ghazy, A. A.-H. El-Asmy and A. M. El-Nokrashy, "Batch Removal of Nickel by Eggshell as a Low Cost Sorbent," *International Journal of Industrial Chemistry*, 2011.
- [36] C. Lu, C. Liu and G. P. Rao, "Comparisons of sorbent cost for the removal of Ni²⁺ from aqueous solution by carbon nanotubes and granular activated carbon," *Journal of Hazardous Materials*, 2007.
- [37] Z.-n. Huang and X.-l. Wang, "Adsorption of Cr(VI) in wastewater using magnetic multi-wall carbon nanotubes," 2015.
- [38] R. Kumar, M. O. Ansari and M. Barakat, "DBSA doped polyaniline/multimultiwalled on nanotubes composite for high efficiency removal of Cr (VI) from aqueous solution," *Elsevier*, 2013.
- [39] K. Kadirvelu, C. F. Brasquet and P. L. Cloirec, "Removal of Cu(II), Pb(II), and Ni(II) by Adsorption onto Activated Carbon Cloths," *Langmuir*, vol. 16, 2000.
- [40] Z. Reddad, C. Gerente, Y. Andres and P. L. Cloirec, "Adsorption of Several Metal Ions onto a Low-Cost Biosorbent: Kinetic and Equilibrium Studies," *Environ. Sci. Technol*, 2002.
- [41] A. H. Hawari and C. N. Mulligan, "Biosorption of lead(II), cadmium(II), copper(II) and nickel(II) by anaerobic granular biomass," *Bioresource Technology*, vol. 97, 2006.
- [42] S. Aslan, A. Polat and U. S. Topcu, "Assessment of the adsorption kinetics, equilibrium and thermodynamics for the potential removal of Ni²⁺ from aqueous solution using waste eggshell," *Journal of Environmental Engineering and Landscape Management*, pp. 221 - 229, 2015.

- [43] W. Konicki, I. Pelech and E. Mijowska, "Removal of Ni²⁺ from aqueous solutions by adsorption onto magnetic multiwalled carbon nanotube nanocomposite," *Polish Journal of Chemical Technology*, vol. 16, 2014.
- [44] A. R. Ipeaiyeda and G. O. Tesi, "Sorption and Desorption Studies on Toxic Metals From Brewery Effluent Using Eggshell as Adsorbent," *Advances in Natural Science*, pp. 15-24, 2014.
- [45] S. Aslan, A. Polat and U. S. Topcu, "Assessment of the adsorption kinetics, equilibrium and thermodynamics for the potential removal of Ni²⁺ from aqueous solution using waste eggshell," *Journal of Environmental Engineering and Landscape Management*, pp. 221-229, 2015.
- [46] B. Pan and B. Xing, "Adsorption Mechanisms of Organic Chemicals on Carbon Nanotubes," *Environmental Science & Technology*, 2009.
- [47] M. Abdullah, F. Iskandar and K. Okuyama, "Simple Fabrication of Carbon Nanotubes from Ethanol using an Ultrasonic Spray Pyrolysis," *PROC. ITB Eng. Science*, vol. 36 B, pp. 125-131, 2004.
- [48] M. Aliabadi, K. Morshedzadeh and H. Soheyli, "Removal of hexavalent chromium from aqueous solution by lignocellulosic solid wastes," *Elsevier*, 2006.
- [49] ., U. Garg, M. Kaur, V. Garg and D. Sud, "Removal of hexavalent chromium from aqueous solution by agricultural waste biomass," *Elsevier*, 2007.
- [50] G. Agarwal, H. Bhuptawat and S. Chaudhari, "Biosorption of aqueous chromium (VI) by *Tamarindus indica* seeds," *Elsevier*, 2006.
- [51] H. Gao, Y. Liu, G. Zeng, W. Xu, T. Li and W. Xia, "Characterization of Cr (VI) removal from aqueous solutions by a surplus agricultural waste-rice straw," *Elsevier*, 2008.
- [52] L. Sun, Z. Yuan, W. Gong, L. Zhang, X. Zili and S. Gongbing, "The mechanism study of trace Cr (VI) removal from water using Fe₀nanorods modified with chitosan in porous anodic alumina," *Elsevier*, 2015.
- [53] M. H. Dehghani, M. M. Taher, A. K. Bajpai and B. Heibati, "Removal of noxious Cr (VI) ions using single-walled carbon nanotubes and multi-walled carbon nanotubes," *Elsevier*, 2015.
- [54] C. Sakulthaew, C. Chokejaroenrat, A. Poapolathep and T. Satapanajaru, "Hexavalent chromium adsorption from aqueous solution using carbon nano-onions (CNOs)," *Elsevier*, 2017.
- [55] Z.-C. Di, Y.-H. Li, Z.-K. Luan and J. Liang, "Adsorption of Chromium(VI) Ions from Water by Carbon Nanotubes," 2004.

- [56] Z.-n. Huang and X.-l. Wang, "Adsorption of Cr(VI) in wastewater using magnetic multi-wall carbon nanotubes," 2015.
- [57] D. Milenkovic and A. Bojic, "Ultrasound-assisted adsorption of 4-dodecylbenzene sulfonate from aqueous solutions by corn cob activated carbon," *Elsevier*, 2013.
- [58] J. Hu, S. Wang, D. Shao and Y. Dong, "Adsorption and Reduction of Chromium(VI) from Aqueous Solution by Multiwalled Carbon Nanotubes," 2009.