

BAB V

KESIMPULAN DAN SARAN

5.1 Kesimpulan

Berdasarkan percobaan yang telah dilakukan, dapat disimpulkan bahwa:

1. Seiring dengan meningkatnya konsentrasi awal zat warna, profil *%removal* zat warna *Congo* merah mengalami penurunan dari 43,25% ke 30,34% (30°C); 56,74% ke 33,67% (40°C); dan 88,77% ke 41,04% (50°C). Hal ini dipengaruhi oleh menurunnya kemampuan muatan positif protein untuk menetralisasi muatan negatif *Congo* merah yang mengalami kenaikan dalam larutan.
2. Pada konsentrasi yang sama, *%removal* zat warna *Congo* merah mengalami peningkatan dari 43,25% ke 88,77% seiring dengan naiknya temperatur koagulasi. Hal ini disebabkan oleh semakin tingginya kontak antara partikel dalam larutan pada temperatur yang lebih tinggi, sehingga flok yang terbentuk menjadi lebih banyak dan meningkatkan *%removal*.
3. Model kinetika *pseudo-second order* merupakan model yang paling cocok untuk menggambarkan proses koagulasi zat warna *Congo* merah dengan ekstrak biji petai cina, dengan nilai konstanta laju (k_2) berada pada rentang $2,1317 - 3,1997 \text{ g/mg.min}$ (30°C); $1,3347 - 3,5101 \text{ g/mg.min}$ (40°C) dan $1,1421 - 3,7192 \text{ g/mg.min}$ (50°C). Model ini menunjukkan bahwa mekanisme adsorpsi-netralisasi muatannya merupakan *chemisorption* yang terjadi karena adanya interaksi antara gugus protein dan *Congo* merah.
4. Model isoterm Langmuir merupakan model yang paling cocok untuk menggambarkan proses koagulasi zat warna *Congo* merah dengan ekstrak biji petai cina, dengan nilai konstanta (K_L) berada pada rentang $0,0491 - 1,6650 \text{ L/mg}$. Model ini menunjukkan bahwa mekanisme adsorpsi-netralisasi terjadi pada permukaan yang homogen dan bersifat monolayer.
5. Hasil termodinamika menunjukkan proses adsorpsi zat warna *Congo* merah pada ekstrak biji petai cina berlangsung secara spontan yang ditunjukkan dengan energi bebas Gibss bernilai negatif, endotermis yang ditunjukkan dengan entalpi bernilai positif (142,96 kJ/mol), serta adanya peningkatan ketidakteraturan interaksi pada permukaan koagulan yang ditunjukkan dengan entropi bernilai positif (0,45 kJ/mol.K).

5.2 Saran

Beberapa saran yang dapat diberikan untuk penelitian lebih lanjut adalah:

1. Perlu dilakukan pengecekan secara berkala pada pH larutan *Congo* merah agar tetap berada di bawah titik isoelektriknya sebelum dilakukan koagulasi

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