

BAB V

KESIMPULAN DAN SARAN

5.1. Kesimpulan

Kegiatan penelitian yang dilakukan terdapat beberapa hal yang dapat disimpulkan. Adapun kesimpulan yang dihasilkan adalah sebagai berikut:

1. Proses sintesis melalui proses HTC dan pirolisis menghasilkan *yield* produk *N-doped hard carbon* sebesar 22,39 % untuk rasio *hydrochar*: NaNH_2 1:1 pada suhu aktivasi 500 °C dan seiring penambahan NaNH_2 serta kenaikan suhu aktivasi, %*yield* akan menurun hingga 13,47 % untuk rasio *hydrochar*: NaNH_2 1:2 pada suhu aktivasi 700 °C. Sedangkan, proses sintesis melalui proses *direct pyrolysis* memiliki *yield* produk *N-doped hard carbon* yang lebih besar yaitu 51,8 % hingga 33,67 % dan memiliki kecenderungan yang sama seperti sintesis proses 2 tahap.
2. Pada hasil analisis SEM, didapati hasil sintesis proses HTC dan pirolisis menghasilkan *hard carbon* dengan struktur *microspheres* dan sedikit pori, sedangkan pada sintesis proses *direct pyrolysis*, struktur yang dihasilkan acak dengan permukaan yang tidak merata dan berpori serta terbentuk *flakes*. Penambahan *sodamide* ini juga berpengaruh terhadap pembentukan struktur *hard carbon*, semakin banyak *sodamide* yang ditambahkan, semakin banyak pori yang terbentuk pada prekursor karbon.
3. Berdasarkan analisis komposisi sampel dengan metode EDS, *N-doping* berhasil dilakukan pada kedua variasi tahap. Metode dengan proses HTC dan pirolisis menghasilkan *N-doped hard carbon* yang lebih tinggi unsur nitrogen dibandingkan dengan proses *direct pyrolysis* dengan % unsur nitrogen sebesar 0,88 % hingga 1,58 %. Pada proses *direct pyrolysis* didapatkan hasil % unsur nitrogen sebesar 0,67 % hingga 0,96 %. Kenaikan suhu dan jumlah penambahan *sodamide* juga berbanding lurus dengan kenaikan unsur nitrogen yang berhasil terdoping.
4. Pada hasil analisis XRD, didapatkan *N-doped hard carbon* hasil penelitian ini memiliki struktur *amorphous* (mencapai 72,57 % amorf) yang lebih dominan dibandingkan struktur kristalin yang memiliki jarak *interlayer* (d_{002}) berkisar

0,363 nm hingga 0,393 nm dan jarak *intercrystallite* (d_{100}) yang berkisar 0,203 nm hingga 0,211 nm. Nilai d_{002} didapatkan dari proses *direct pyrolysis* sedikit lebih besar dibandingkan dengan proses yang melewati HTC terlebih dahulu, dengan kecenderungan kenaikan suhu akan meningkatkan nilai dari d_{002} .

5. Produk *N-doped hard carbon* yang melewati karbonisasi dua tahap (HTC + pirolisis) lebih baik dibandingkan dengan proses karbonisasi satu tahap (*direct pyrolysis*) karena memiliki kandungan % unsur nitrogen yang lebih banyak.

5.2. Saran

Demi kepentingan pengembangan penelitian selanjutnya, penelitian ini memiliki beberapa saran yang dapat dipertimbangkan sebagai berikut:

1. Analisis hasil penelitian ini dapat dikembangkan dengan melakukan uji karakteristik elektrokimia seperti *electrochemical performance* dan *cycling performance*, sehingga dapat diketahui bahwa hard carbon yang dihasilkan menunjukkan kecocokkan untuk penggunaan *sodium ion batteries*.
2. Selain analisis SEM-EDS dan XRD, juga dapat dilakukan analisis XPS dan FTIR untuk mengidentifikasi komposisi dan gugus fungsi nitrogen dengan lebih akurat.
3. Analisis BET dapat dilakukan untuk mengetahui *surface area* dan distribusi pori dari produk *N-doped hard carbon* yang dihasilkan.
4. Variasi temperatur aktivasi dapat dilakukan lebih lanjut sehingga dapat diketahui variasi temperatur aktivasi yang lebih berpengaruh terhadap pembentukan *hard carbon* yang cocok untuk *sodium ion batteries*.

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