

BAB 5

KESIMPULAN DAN SARAN

5.1. Kesimpulan

Kegiatan penelitian yang dilakukan menghasilkan beberapa kesimpulan. Adapun kesimpulan yang dihasilkan adalah sebagai berikut.

1. Proses *direct carbonization* menghasilkan *yield* produk *N-doped hard carbon* dengan *yield* 22,67% untuk rasio pati:urea 1:1, serta 27,30% untuk rasio massa pati:urea 1:2. Sedangkan, proses sintesis melalui HTC memiliki *yield* produk 7,39% untuk rasio pati:urea 1:1, dan 6,70% untuk rasio pati:urea 1:2.
2. Berdasarkan analisis SEM, produk dihasilkan dengan adanya proses HTC memiliki morfologi berupa *microsphere*, sedangkan produk dari *direct carbonization* memiliki morfologi dengan bentuk *flakes*. Modifikasi berupa penambahan urea menghasilkan produk *N-doped hard carbon* dengan adanya aglomerasi, serta diameter *microsphere* yang lebih kecil.
3. Berdasarkan analisis komposisi sampel dengan metode EDS, *N-doping* berhasil dilakukan pada sintesis melalui proses HTC yang menghasilkan produk *N-doped hard carbon*. Selanjutnya, perubahan rasio massa pati terhadap urea dari 1:1 menjadi 1:2 meningkatkan komposisi nitrogen pada *N-doped hard carbon* dari 2,12% menjadi 3,08%. Sebaliknya, proses *direct carbonization* tidak memberikan *N-doping* pada produk *hard carbon*. Kemudian, sampel *N-doped hard carbon* dari proses HTC juga memiliki kandungan karbon yang lebih tinggi yaitu 90,12% dan 90,19% dibandingkan proses *direct carbonization* sebesar 87,29% dan 86,43% untuk rasio massa pati:urea 1:1 dan 1:2 secara berurutan.
4. Berdasarkan analisis XRD, produk *N-doped hard carbon* dari proses HTC dan *direct carbonization* memiliki *interlayer spacing* di rentang 0,362 hingga 0,382 nm, dimana *interlayer spacing* pada proses HTC sedikit lebih kecil dibandingkan *direct carbonization*. Selain itu, analisa kristalinitas XRD juga menunjukkan bahwa produk *N-doped hard carbon* dari proses HTC juga memiliki komposisi *amorphous* yang lebih besar (mencapai 72,4% *amorphous*) dibandingkan dari proses *direct carbonization* (hanya mencapai 67,4% *amorphous*).

5. Berdasarkan analisis Raman *spectroscopy*, produk *N-doped hard carbon* dari proses HTC memiliki nilai ID/IG sebesar 1,26 dan 1,31 pada rasio massa urea:pati senilai 1:1 dan 1:2. Sedangkan, nilai ID/IG pada produk *direct carbonization* adalah 1,08 dan 1,13 pada rasio 1:1 dan 1:2. Akibatnya, dapat disimpulkan bahwa proses sintesis dengan HTC dan penambahan jumlah urea meningkatkan sifat *amorphous* dari produk *N-doped hard carbon*.

5.2. Saran

Untuk pengembangan kegiatan penelitian selanjutnya, terdapat beberapa saran yang dapat dipertimbangkan sebagai berikut.

1. Eksplorasi lebih jauh mengenai metode modifikasi pati untuk penambahan unsur nitrogen yang lebih efektif.
2. Analisis SEM (EDS) dan XRD terhadap sampel-sampel bahan baku pati, pati termodifikasi dan *hydrochar*. Selain itu, produk *N-doped hard carbon* juga dapat dianalisis dengan metode XPS dan FTIR untuk mengidentifikasi komposisi dan gugus fungsi nitrogen dengan lebih akurat.
3. Pengujian performa elektrokimia variasi sampel *N-doped hard carbon* sebagai material anoda *sodium-ion batteries* (SIB).

DAFTAR PUSTAKA

- Ahmad, W. R. W., Mamat, M. H., Zoolfakar, A. S., Khusaimi, Z., & Rusop, M. (2017). A review on hematite α -Fe₂O₃ focusing on nanostructures, synthesis methods and applications. *Proceedings - 14th IEEE Student Conference on Research and Development: Advancing Technology for Humanity, SCOReD 2016*, 2–7. <https://doi.org/10.1109/SCORED.2016.7810090>
- Alcázar-Alay, S. C., & Meireles, M. A. A. (2015). Physicochemical properties, modifications and applications of starches from different botanical sources. *Food Science and Technology*, 35(2), 215–236. <https://doi.org/10.1590/1678-457X.6749>
- Alexandrova, A. N., & Jorgensen, W. L. (2007). Why urea eliminates ammonia rather than hydrolyzes in aqueous solution. *Journal of Physical Chemistry B*, 111(4), 720–730. <https://doi.org/10.1021/jp066478s>
- Alhnidi, M. J., Körner, P., Wüst, D., Pfersich, J., & Kruse, A. (2020). Nitrogen-Containing Hydrochar: The Influence of Nitrogen-Containing Compounds on the Hydrochar Formation. *ChemistryOpen*, 9(8), 864–873. <https://doi.org/10.1002/open.202000148>
- Alvin, S., Yoon, D., Chandra, C., Susanti, R. F., Chang, W., Ryu, C., & Kim, J. (2019). Extended flat voltage profile of hard carbon synthesized using a two-step carbonization approach as an anode in sodium ion batteries. *Journal of Power Sources*, 430(April), 157–168. <https://doi.org/10.1016/j.jpowsour.2019.05.013>
- Andreas, A., Burak, A., Demir, E., & Demir, R. (2019). Utilization of The Indonesian 's Spent Tea Leaves as Promising Porous Hard Carbon Precursors for Anode Materials in Sodium Ion Batteries. *Waste and Biomass Valorization*, 0(0), 0. <https://doi.org/10.1007/s12649-019-00624-x>
- Anggarini, D., Hidayat, N., & Mulyadi, A. F. (2016). Canna Edulis Starch as the Raw Material of Edible coating and It's Application on the Storage of Anna Apples (*Malus sylvestris*) (The Study of Canna Edulis Starch and Glycerol Concentrate). *Industria : Jurnal Teknologi dan Manajemen Agroindustri*, 5(1), 1–8.
- Ban, L. L., Crawford, D., & Marsh, H. (1975). Lattice-resolution electron microscopy in structural studies of non-graphitizing carbons from polyvinylidene chloride (PVDC). *Journal of Applied Crystallography*, 8(4), 415–420. <https://doi.org/10.1107/s0021889875010904>
- Barron, Andrew R., & Raja, P. M. V. (2012). *Physical Methods in Chemistry and Nano Science* (Andrew R. Barron (ed.)). Rice University, Houston, Texas (Connexions). <http://cnx.org/content/col10699/1.14/>
- Beda, A., Taberna, P. L., Simon, P., & Matei Ghimbeu, C. (2018). Hard carbons derived from green phenolic resins for Na-ion batteries. *Carbon*, 139, 248–257. <https://doi.org/10.1016/j.carbon.2018.06.036>
- Berge, N. D., Ro, K. S., Mao, J., Flora, J. R. V., Chappell, M. A., & Bae, S. (2011). Hydrothermal carbonization of municipal waste streams. *Environmental Science and Technology*, 45(13), 5696–5703. <https://doi.org/10.1021/es2004528>

- Buiel, E. R., George, A. E., & Dahn, J. R. (1999). Model of micropore closure in hard carbon prepared from sucrose. *Carbon*, *37*(9), 1399–1407. [https://doi.org/10.1016/S0008-6223\(98\)00335-2](https://doi.org/10.1016/S0008-6223(98)00335-2)
- Cao, Y., Xiao, L., Sushko, M. L., Wang, W., Schwenzer, B., Xiao, J., Nie, Z., Saraf, L. V., Yang, Z., & Liu, J. (2013). Sodium ion insertion in hollow carbon nanowires for battery applications. *Nano Letters*, *4*(7), 1870. <https://doi.org/10.1021/nl400998t>
- Chen, H., Cong, T. N., Yang, W., Tan, C., Li, Y., & Ding, Y. (2009). Progress in electrical energy storage system: A critical review. *Progress in Natural Science*, *19*(3), 291–312. <https://doi.org/10.1016/j.pnsc.2008.07.014>
- Cresswell, S. L., & Haswell, S. J. (2001). Microwave ovens-out of the kitchen. *Journal of Chemical Education*, *78*(7), 900–904. <https://doi.org/10.1021/ed078p900>
- Dahn, J. R., Zheng, T., Liu, Y., & Xue, J. S. (1995). Mechanisms for lithium insertion in carbonaceous materials. *Science*, *270*(5236), 590–593. <https://doi.org/10.1126/science.270.5236.590>
- Demir-Cakan, R., Baccile, N., Antonietti, M., & Titirici, M. M. (2009). Carboxylate-rich carbonaceous materials via one-step hydrothermal carbonization of glucose in the presence of acrylic acid. *Chemistry of Materials*, *21*(3), 484–490. <https://doi.org/10.1021/cm802141h>
- Dou, X., Hasa, I., Hekmatfar, M., Diemant, T., & J, R. (2017). Pectin , Hemicellulose , or Lignin ? Impact of the Biowaste Source on the Performance of Hard Carbons for Sodium-Ion Batteries. *ChemSusChem*, *10*(12), 2668–2676. <https://doi.org/10.1002/cssc.201700628>
- Dou, X., Hasa, I., Saurel, D., Vaalma, C., Wu, L., Buchholz, D., Bresser, D., Komaba, S., & Passerini, S. (2019). Hard carbons for sodium-ion batteries : Structure , analysis , sustainability , and electrochemistry. *Materials Today*, *23*(March), 87–104. <https://doi.org/10.1016/j.mattod.2018.12.040>
- Dutrow, B. L. (Louisiana S. U., & Clark, C. M. (Eastern M. U. (2020). *X-ray Powder Diffraction (XRD)*. https://serc.carleton.edu/research_education/geochemsheets/techniques/XRD.html
- El-thalouth, I. A., El-Kashouti, M. A., & Hebeish, A. (1981). Modification of Rice Starch through Thermal Treatment with Urea. *Starch- Stärke*, *33*(9), 306–310.
- Engelke, S. (2013). Current and future sodium-ion battery research. *Int. Journal of Energy Storage-Draft*, *1*(March 2013), 1–7.
- Ford, B. J., Joy, D. C., & Bradbury, S. (2019). *Scanning electron microscope*. Encyclopædia Britannica; Encyclopædia Britannica. <https://www.britannica.com/technology/scanning-electron-microscope/images-videos#/media/1/526571/110970>
- Franklin, R. E. (1951). Crystallite growth in graphitizing and non-graphitizing carbons. *Proceedings of the Royal Society of London. Series A. Mathematical and Physical Sciences*, *209*(1097), 196–218. <https://doi.org/10.1098/rspa.1951.0197>

- Fu, Y., Wei, Q., Zhang, G., & Sun, S. (2018). Advanced Phosphorus-Based Materials for Lithium / Sodium-Ion Batteries: Recent Developments and Future Perspectives. *Advanced Energy Materials*, 8(13), 1703058. <https://doi.org/10.1002/aenm.201702849>
- Funke, A., & Ziegler, F. (2012). Hydrothermal carbonization of biomass: A summary and discussion of chemical mechanisms for process engineering. *Biofuels, Bioproducts and Biorefining*, 6(3), 246–256. <https://doi.org/10.1002/bbb>
- Gaddam, R. R., Niaei, A. H. F., Hankel, M., Searles, D. J., Kumar, N. A., & Zhao, X. S. (2017). Capacitance-enhanced sodium-ion storage in nitrogen-rich hard carbon. *Journal of Materials Chemistry A*, 5(42), 22186–22192. <https://doi.org/10.1039/C7TA06754B>
- Geddes, R. (1969). Starch biosynthesis. *Quarterly Reviews, Chemical Society*, 23(1), 57–72. <https://doi.org/10.1039/qr9692300057>
- Goodge, J. (University of M.-D. (2017). *Energy-Dispersive X-Ray Spectroscopy (EDS)*. https://serc.carleton.edu/research_education/geochemsheets/eds.html
- Harmayani, E. (Universitas G. M., Murdiati, A. (Universitas G. M., & Griyaningsih (Universitas Gadjah Mada). (2011). Characterization of Edible Canna Starch (*Canna edulis*) and Its Application as Ingredient for Cookies and Cendol. *Agritech*, 31(4), 297–304. <https://doi.org/https://doi.org/10.22146/agritech.9637>
- Harris, P. J. F. (1997). Structure of non-graphitising carbons. *Journal International Materials Reviews*, 42(5), 206–218.
- Hasanah, F., & Hasrini, R. F. (2018). Pemanfaatan Ganyong (*Canna edulis* KERR) sebagai Bahan Baku Sohun dan Analisis Kualitasnya. *Warta IHP (Industri Hasil Pertanian)*, 35(2), 99–105.
- Hebeish, A., El-thalouth, I. A., & El-Kashouti, M. A. (1981). Chemical Modification of Starch II. Cyanoethylation. *Journal of Applied Polymer Science*, 26(1), 171–176.
- Hebeish, A., Refai, R., & Ragheb, A. (1991). Factors Affecting the Technological Properties of Starch Carbamate *. *Starch-Stärke*, 43(7), 273–280.
- Henry, D. (Louisiana S. U., Eby, N. (University of M.-L., Goodge, J. (University of M.-D., & Mogk, D. (Montana S. U. (2016). *X-ray reflection in accordance with Bragg's Law*. https://serc.carleton.edu/research_education/geochemsheets/BraggsLaw.html
- Hou, H., Yu, C., Liu, X., Yao, Y., Dai, Z., & Li, D. (2019). *The effect of carbonization temperature of waste cigarette butts on Na - storage capacity of N - doped hard carbon anode. Harris 2013*.
- Hu, B., Wang, K., Wu, L., Yu, S. H., Antonietti, M., & Titirici, M. M. (2010). Engineering carbon materials from the hydrothermal carbonization process of biomass. *Advanced Materials*, 22(7), 813–828. <https://doi.org/10.1002/adma.200902812>
- Hu, M., Yang, L., Zhou, K., Zhou, C., Huang, Z., Kang, F., & Lv, R. (2017). Enhanced sodium-ion storage of nitrogen-rich hard carbon by NaCl intercalation. *Carbon*, 122, 680–686. <https://doi.org/10.1016/j.carbon.2017.05.003>

- Kalvin. (2020). *Sintesis hard carbon dari pati sagu melalui hydrothermal carbonization (HTC) dan aktivasi heat treatment temperature (HTT)*. Universitas Katolik Parahyangan.
- Khalil, M., Farag, S., Mostafa, K. M., & Hebeish, A. (1994). Some Studies on Starch Carbamate. *Starch-Stärke*, 46(8), 312–316.
- Khosravi, M., Bashirpour, N., & Nematpour, F. (2014). Synthesis of hard carbon as anode material for lithium ion battery. *Advanced Materials Research*, 829, 922–926. <https://doi.org/10.4028/www.scientific.net/AMR.829.922>
- Kirk-Othmer. (1998). Starch. In *Encyclopedia of Chemical Technology* (4th editio, hal. 340–349).
- Koswara, S. (2009). Teknologi Modifikasi Pati. In *Ebook Pangan*. Teknik Pangan, Universitas Muhammadiyah Semarang. <https://doi.org/10.1016/B978-1-85573-731-0.50013-X>
- Kristianto, H., Arie, A. A., Susanti, R. F., Halim, M., & Lee, J. K. (2016). The effect of activated carbon support surface modification on characteristics of carbon nanospheres prepared by deposition precipitation of Fe-catalyst. *IOP Conference Series: Materials Science and Engineering*, 162(1). <https://doi.org/10.1088/1757-899X/162/1/012034>
- Lewicka, K., Siemion, P., & Kurcok, P. (2015). Chemical modifications of starch: Microwave effect. *International Journal of Polymer Science*, 2015, 10. <https://doi.org/10.1155/2015/867697>
- Li, R., & Shahbazi, A. (2015). A Review of Hydrothermal Carbonization of Carbohydrates for Carbon Spheres Preparation. *Trends in Renewable Energy*, 1(1), 43–56. <https://doi.org/10.17737/tre.2015.1.1.009>
- Li, W., Chen, M., & Wang, C. (2011). Spherical hard carbon prepared from potato starch using as anode material for Li-ion batteries. *Materials Letters*, 65(23–24), 3368–3370. <https://doi.org/10.1016/j.matlet.2011.07.072>
- Li, Z., Bommier, C., Chong, Z. Sen, Jian, Z., Surta, T. W., Wang, X., Xing, Z., Neuefeind, J. C., Stickle, W. F., Dolgos, M., Greaney, P. A., & Ji, X. (2017). Mechanism of Na-Ion Storage in Hard Carbon Anodes Revealed by Heteroatom Doping. *Advanced Energy Materials*, 7(18), 1-10 (1602894). <https://doi.org/10.1002/aenm.201602894>
- Liang, J. L., Liu, Y. H., & Zhang, J. (2011). Effect of solution pH on the carbon microsphere synthesized by hydrothermal carbonization. *Procedia Environmental Sciences*, 11(PART C), 1322–1327. <https://doi.org/10.1016/j.proenv.2011.12.198>
- Libra, J. A., Ro, K. S., Kammann, C., Funke, A., Berge, N. D., Neubauer, Y., Titirici, M. M., Fühner, C., Bens, O., Kern, J., & Emmerich, K. H. (2011). Hydrothermal carbonization of biomass residuals: A comparative review of the chemistry, processes and applications of wet and dry pyrolysis. *Biofuels*, 2(1), 71–106. <https://doi.org/10.4155/bfs.10.81>
- Marsh, H., Edwards, I., Menendez, R., Rand, B., West, S., Hosty, A., Kuo, K., McEnaney, B., Timothy, M., Johnson, D., Patrick, J., Clarke, D., Crelling, J., & Gray, R. (1991). Introduction to carbon science. In H. Marsh (Ed.), *The Chemical Engineering Journal* (Vol. 46, Nomor 1). Butterworth & Co. [https://doi.org/10.1016/0300-9467\(91\)80007-j](https://doi.org/10.1016/0300-9467(91)80007-j)

- Morikawa, Y., Nishimura, S. ichi, Hashimoto, R. ichi, Ohnuma, M., & Yamada, A. (2019). Mechanism of Sodium Storage in Hard Carbon: An X-Ray Scattering Analysis. *Advanced Energy Materials*, 10(3), 1-9 (1903176). <https://doi.org/10.1002/aenm.201903176>
- Moriwake, H., Kuwabara, A., Fisher, C. A. J., & Ikuhara, Y. (2017). Why is sodium-intercalated graphite unstable? *RSC Advances*, 7, 36550–36554. <https://doi.org/10.1039/C7RA06777A>
- Ngidi, N. P. D., Ollengo, M. A., & Nyamori, V. O. (2019). Precursors on the Physicochemical , Optical , Nitrogen-Doped Reduced Graphene Oxide. *Materials*, 12(20), 3376–3401.
- Pangesthi, L. T. (2009). Pemanfaatan Pati Ganyong (*Canna Edulis*) Pada Pembuatan Mie Segar Sebagai Upaya Penganekaragaman Pangan Non Beras. *Jurnal Media, Pendidikan, Gizi dan Kuliner*, 1(1).
- Piyachomkwan, K., Chotineeranat, S., Kijkhunasatian, C., Tonwitawat, R., Prammanee, S., Oates, C. G., & Sriroth, K. (2002). Edible canna (*Canna edulis*) as a complementary starch source to cassava for the starch industry. *Industrial Crops and Products*, 16(1), 11–21. [https://doi.org/10.1016/S0926-6690\(02\)00003-1](https://doi.org/10.1016/S0926-6690(02)00003-1)
- Putri, C. C. (2021). *Sintesis N-doped Hard Carbon dari Pati Sagu Termodifikasi dengan Proses Karbonisasi Hidrotermal dan Aktivasi*. Universitas Katolik Parahyangan.
- Raguin, A., & Ebenhöf, O. (2017). Design starch: Stochastic modeling of starch granule biogenesis. *Biochemical Society Transactions*, 45(4), 885–893. <https://doi.org/10.1042/BST20160407>
- Ratchahat, S., Viriya-Empikul, N., Faungnawakij, K., Charinpanitkul, T., & Soottitantawat, A. (2010). Synthesis of Carbon Microspheres from Starch by Hydrothermal Process. *Sci. J. UBU*, 1(2), 40–45.
- Reza, M. T., Andert, J., Wirth, B., Busch, D., Pielert, J., Lynam, J. G., & Mumme, J. (2014). Hydrothermal Carbonization of Biomass for Energy and Crop Production. *Applied Bioenergy*, 1(1), 11–29. <https://doi.org/10.2478/apbi-2014-0001>
- Reza, M. T., Rottler, E., Herklotz, L., & Wirth, B. (2015). Hydrothermal carbonization (HTC) of wheat straw: Influence of feedwater pH prepared by acetic acid and potassium hydroxide. *Bioresource Technology*, 182(April 2015), 336–344. <https://doi.org/10.1016/j.biortech.2015.02.024>
- Richardson, S., & Gorton, L. (2003). Characterisation of the substituent distribution in starch and cellulose derivatives. *Analytica Chimica Acta*, 497(1–2), 27–65. <https://doi.org/10.1016/j.aca.2003.08.005>
- Salinas-Torres, D., Shiraishi, S., Morallo, E., & Cazorla-Amoros, D. (2014). Improvement of carbon materials performance by nitrogen functional groups in electrochemical capacitors in organic electrolyte at severe conditions. *Carbon*, 82, 205–213. <https://doi.org/10.1016/j.carbon.2014.10.064>
- Santacruz, S. (2004). *Characterisation of starches isolated from Arracacha xanthorrhiza, Canna edulis and Oxalis tuberosa and extracted from potato leaf* [Swedish University of Agricultural Sciences]. <http://pub.epsilon.slu.se/658/1/Agraria486.pdf>

- Schaber, P. M., Colson, J., Higgins, S., Thielen, D., Anspach, B., & Brauer, J. (2004). Thermal decomposition (pyrolysis) of urea in an open reaction vessel. *Thermochimica Acta*, 424(1–2), 131–142. <https://doi.org/10.1016/j.tca.2004.05.018>
- Sevilla, M., & Fuertes, A. B. (2009). The production of carbon materials by hydrothermal carbonization of cellulose. *Carbon*, 47(9), 2281–2289. <https://doi.org/10.1016/j.carbon.2009.04.026>
- Sevilla, Marta, & Fuertes, A. B. (2009). Chemical and structural properties of carbonaceous products obtained by hydrothermal carbonization of saccharides. *Chemistry - A European Journal*, 15(16), 4195–4203. <https://doi.org/10.1002/chem.200802097>
- Shinn, J. H. (1984). From coal to single-stage and two-stage products: A reactive model of coal structure. *Fuel*, 63(9), 1187–1196. [https://doi.org/10.1016/0016-2361\(84\)90422-8](https://doi.org/10.1016/0016-2361(84)90422-8)
- Siemion, P., Jabłońska, J., Kapuśniak, J., & Koziół, J. J. (2004). Solid state reactions of potato starch with urea and biuret. *Journal of Polymers and the Environment*, 12(4), 247–255. <https://doi.org/10.1007/s10924-004-8152-2>
- Siemion, P., Kapusniak, J., & Koziół, J. J. (2005a). Solid-state thermal reactions of starch with semicarbazide hydrochloride . Cationic starches of a new generation. *Carbohydrate Polymers*, 62(2), 182–186. <https://doi.org/10.1016/j.carbpol.2005.07.025>
- Siemion, P., Kapusniak, J., & Koziół, J. J. (2005b). Thermally Induced Reaction of Potato Starch with Thiourea. *Journal of Polymers and the Environment*, 13(1), 19–27. <https://doi.org/10.1007/s10924-004-1212-9>
- Silva, V. A. da, & Rezende, M. C. (2018). Effect of the Morphology and Structure on the Microwave Absorbing Properties of Multiwalled Carbon Nanotube Filled Epoxy Resin Nanocomposites. *Materials Research*, 21(5). <https://doi.org/10.1590/1980-5373-mr-2017-0977>
- Stevens, D. A., & Dahn, J. R. (2000). High Capacity Anode Materials for Rechargeable Sodium-Ion Batteries. *Journal of The Electrochemical Society*, 147(4), 1271.
- Stevens, D. A., & Dahn, J. R. (2001). The Mechanisms of Lithium and Sodium Insertion in Carbon Materials. *Journal of The Electrochemical Society*, 148(8), A803. <https://doi.org/10.1149/1.1379565>
- Sun, X., & Li, Y. (2004). Colloidal Carbon Spheres and Their Core/Shell Structures with Noble-Metal Nanoparticles. *Angewandte Chemie*, 116(5), 607–611. <https://doi.org/10.1002/ange.200352386>
- Swapp, S. (University of W. (2017). *Scanning Electron Microscopy (SEM)*. https://serc.carleton.edu/research_education/geochemsheets/techniques/SEM.html
- Tanuwijaya, K. H. (2020). *Sintesis hard carbon berbahan dasar pati ganyong sebagai anoda baterai sodium*. Universitas Katolik Parahyangan.
- Thitipraphunkul, K., Uttapap, D., Piyachomkwan, K., & Takeda, Y. (2003a). A comparative study of edible canna (*Canna edulis*) starch from different cultivars. Part I: Chemical composition and physicochemical properties. *Carbohydrate Polymers*, 53(3), 317–324. [https://doi.org/10.1016/S0144-8617\(03\)00081-X](https://doi.org/10.1016/S0144-8617(03)00081-X)

- Thitipraphunkul, K., Uttapap, D., Piyachomkwan, K., & Takeda, Y. (2003b). A comparative study of edible canna (*Canna edulis*) starch from different cultivars. Part II. Molecular structure of amylose and amylopectin. *Carbohydrate Polymers*, *54*(4), 489–498. <https://doi.org/10.1016/j.carbpol.2003.08.003>
- Titirici, M.-M. (2013). Sustainable Carbon Materials from Hydrothermal Processes. In M. M. Titirici (Ed.), *Sustainable Carbon Materials from Hydrothermal Processes*. John Wiley & Sons, Ltd. <https://doi.org/10.1002/9781118622179>
- Tsai, P. C., Chung, S. C., Lin, S. K., & Yamada, A. (2015). Ab initio study of sodium intercalation into disordered carbon. *Journal of Materials Chemistry A*, *3*(18), 9763–9768. <https://doi.org/10.1039/c5ta01443c>
- Vignarooban, K., Kushagra, R., Elango, A., & Badami, P. (2016). Current trends and future challenges of electrolytes for sodium-ion batteries. *International Journal of Hydrogen Energy*, *41*(4), 2829–2846. <https://doi.org/10.1016/j.ijhydene.2015.12.090>
- Vollmer, M. (2004). Physics of the microwave oven. *Physics Education*, *39*(1), 74–81. <https://doi.org/10.1088/0031-9120/39/1/006>
- Wang, T., Zhai, Y., Zhu, Y., Li, C., & Zeng, G. (2018). A review of the hydrothermal carbonization of biomass waste for hydrochar formation: Process conditions, fundamentals, and physicochemical properties. *Renewable and Sustainable Energy Reviews*, *90*(February), 223–247. <https://doi.org/10.1016/j.rser.2018.03.071>
- Wang, Z., Qie, L., Yuan, L., Zhang, W., Hu, X., & Huang, Y. (2013). Functionalized N-doped interconnected carbon nanofibers as an anode material for sodium-ion storage with excellent performance. *Carbon*, *55*(April), 328–334. <https://doi.org/10.1016/j.carbon.2012.12.072>
- Wen, Y., He, K., Zhu, Y., Han, F., Xu, Y., Matsuda, I., Ishii, Y., Cumings, J., & Wang, C. (2014). Expanded graphite as superior anode for sodium-ion batteries. *Nature Communications*, *5*(May), 1–10. <https://doi.org/10.1038/ncomms5033>
- Yanagida, K., Yanai, A., KKida, Y., Funahashi, A., Nohma, T., & Yonezu, I. (2002). Carbon hybrids graphite-hard carbon and graphite-coke as negative electrode materials for lithium secondary batteries charge/discharge characteristics. *Journal of The Electrochemical Society*, *149*(7), A804–A807.
- Yang, H., Gong, M., Hu, J., Liu, B., Chen, Y., Xiao, J., Li, S., Dong, Z., & Chen, H. (2020). Cellulose Pyrolysis Mechanism Based on Functional Group Evolutions by Two-Dimensional Perturbation Correlation Infrared Spectroscopy. *Energy & Fuels*, *34*(3), 3412–3421. <https://doi.org/10.1021/acs.energyfuels.0c00134>
- Zhang, J., & Wang, Z. W. (2009). Optimization of reaction conditions for resistant *Canna edulis* Ker starch phosphorylation and its structural characterization. *Industrial Crops and Products*, *30*(1), 105–113. <https://doi.org/10.1016/j.indcrop.2009.02.006>
- Zhang, S., Zhu, X., Zhou, S., Shang, H., Luo, J., & Tsang, D. C. W. (2019). Hydrothermal Carbonization for Hydrochar Production and Its Application. In *Biochar from Biomass and Waste* (hal. 275–294). Elsevier Inc. <https://doi.org/10.1016/b978-0-12-811729-3.00015-7>

- Zhao, L., Baccile, N., Gross, S., Zhang, Y., Wei, W., Sun, Y., Antonietti, M., & Titirici, M. (2010). Sustainable nitrogen-doped carbonaceous materials from biomass derivatives. *Carbon*, 48(13), 3778–3787. <https://doi.org/10.1016/j.carbon.2010.06.040>
- Zheng, P., Liu, T., Zhang, J., Zhang, L., Liu, Y., Huang, J., & Guo, S. (2015). Sweet potato-derived carbon nanoparticles as anode for lithium ion battery. *RSC Advances*, 5(51), 40737–40741. <https://doi.org/10.1039/C5RA03482E>
- Zhu, X., Jiang, X., Liu, X., Xiao, L., & Cao, Y. (2017). A green route to synthesize low-cost and high-performance hard carbon as promising sodium-ion battery anodes from sorghum stalk waste. *Green Energy and Environment*, 2(3), 310–315. <https://doi.org/10.1016/j.gee.2017.05.004>