

BAB V

KESIMPULAN DAN SARAN

5.1 Kesimpulan

Dari penelitian yang telah dilakukan, diperoleh beberapa kesimpulan sebagai berikut:

1. Variasi suhu karbonisasi hidrotermal mempengaruhi *yield hydrochar* yaitu menyebabkan terjadinya penurunan *yield* apabila terjadi peningkatan suhu karbonisasi.
2. *Yield* karbon aktif dipengaruhi oleh penambahan urea dan suhu karbonisasi hidrotermal. Penambahan urea pada metode karbonisasi hidrotermal dan metode pirolisis menyebabkan terjadinya penurunan *yield* karbon aktif. Selain itu, peningkatan suhu karbonisasi hidrotermal menyebabkan terjadinya peningkatan *yield* karbon aktif.
3. Melalui karakterisasi BET, diketahui penambahan urea pada metode karbonisasi hidrotermal dan pirolisis serta peningkatan suhu karbonisasi menyebabkan terjadinya peningkatan luas permukaan, peningkatan total volume pori, peningkatan volume mikropori dan penurunan diameter pori rata-rata.
4. Melalui karakterisasi SEM, diketahui penambahan urea dan peningkatan suhu pada metode karbonisasi hidrotermal menyebabkan semakin banyak terbentuknya pori. Selain itu, penambahan urea juga menyebabkan terjadinya pembentukan struktur yang lebih tipis.
5. Melalui karakterisasi FTIR dan SEM-EDX, diketahui karbon aktif memiliki OFG yang lebih rendah daripada *hydrochar* yang dapat dilihat dari intensitas serapannya dan komposisi komponennya. Komposisi nitrogen dipengaruhi oleh suhu karbonisasi dan proses aktivasi dimana mengalami penurunan ketika suhu karbonisasi meningkat dan menurun juga setelah proses aktivasi berlangsung. Komposisi nitrogen yang paling tinggi diperoleh oleh sampel metode pirolisis.
6. Melalui karakterisasi XRD, diketahui seluruh sampel karbon aktif yang dihasilkan memiliki struktur yang *amorphous*. Struktur dengan %*amorphous* yang paling tinggi dimiliki oleh sampel ACH-225 yaitu sampel metode karbonisasi hidrotermal suhu 225°C tanpa penambahan urea.

5.2 Saran

Dari penelitian yang telah dilakukan, diperoleh beberapa saran yang dapat dipertimbangkan:

1. Perlu dilakukan variasi rasio penambahan urea sehingga dapat diketahui kemungkinan rasio yang lebih baik daripada yang telah diteliti.
2. Perlu dilakukan variasi temperatur pada tahap aktivasi kimia sehingga dapat diperoleh temperatur optimum *N-doped* karbon aktif.
3. Perlu dilakukan karakterisasi XPS (*X-Ray Photoelectron Spectroscopy*) untuk mengetahui kandungan gugus nitrogen dari *N-doped hydrochar* atau *char* dan *N-doped* karbon aktif.
4. Perlu dilakukan karakterisasi elektromia sehingga dapat diketahui kinerja karbon aktif yang paling baik untuk elektroda penyimpan energi.

DAFTAR PUSTAKA

1. Holzinger, C., Robinson, C., and Pelletier, T., *Lux Research - Global Energy Storage Market Forecast*. 2019.
2. Zhai, Y., Dou, Y., Zhao, D., Fulvio, P.F., Mayes, R.T., and Dai, S., *Carbon materials for chemical capacitive energy storage*. Adv Mater, 2011. **23**(42): p. 4828-50.
3. Wang, H., Zhu, C., Chao, D., Yan, Q., and Fan, H.J., *Nonaqueous Hybrid Lithium-Ion and Sodium-Ion Capacitors*. Adv Mater, 2017. **29**(46).
4. Hao, L., Li, X., and Zhi, L., *Carbonaceous electrode materials for supercapacitors*. Adv Mater, 2013. **25**(28): p. 3899-904.
5. Wang, X., Liu, L., and Niu, Z., *Carbon-based materials for lithium-ion capacitors*. Materials Chemistry Frontiers, 2019. **3**(7): p. 1265-1279.
6. Gaspard, S., Altenor, S., Dawson, E.A., Barnes, P.A., and Ouensanga, A., *Activated carbon from vetiver roots: gas and liquid adsorption studies*. J Hazard Mater, 2007. **144**(1-2): p. 73-81.
7. Hady, D., Susanti, R.F., and Arie, A.A., *Preparation of Activated Carbons Originated from Orange Peel Waste by Subcritical H₂O Activation Method*. IOP Conference Series: Materials Science and Engineering, 2020. **742**.
8. Susanti, R.F., Arie, A.A., Kristianto, H., Erico, M., Kevin, G. and Devianto, H., *Activated carbon from citric acid catalyzed hydrothermal carbonization and chemical activation of salacca peel as potential electrode for lithium ion capacitor's cathode*. Ionics, 2019. **25**(8): p. 3915-3925.
9. Chandra, T.C., Mirna, M.M., Sudaryanto, Y., and Ismadji, S., *Adsorption of basic dye onto activated carbon prepared from durian shell: Studies of adsorption equilibrium and kinetics*. Chemical Engineering Journal, 2007. **127**(1-3): p. 121-129.
10. Grاتuito, M.K., Panyathanmaporn, T., Chumnanklang, R.A., Sirinuntawittaya, N., and Dutta, A., *Production of activated carbon from coconut shell: optimization using response surface methodology*. Bioresour Technol, 2008. **99**(11): p. 4887-95.
11. Thwaites, C., *Haitian Vetiver: Uprooted?* 2010. Perfume Flavor. **35**(5): p. 22-23.
12. Rusli, D.M.S., *Sukses Memproduksi Minyak Atsiri*. Agro Media Pustaka. 2010.
13. Anonim. *Minyak Akar Wangi*. 2017 28 Mei 2020; Available from: <https://www.garutkab.go.id/page/minyak-akar-wangi>.

14. Bandosz, T.J., *Activated Carbon Surfaces in Environmental Remediation*. 2006. Elsevier.
15. Kalderis, D., Kotti, M.S., Méndez, A., and Gascó, G., *Characterization of hydrochars produced by hydrothermal carbonization of rice husk*. Solid Earth, 2014. **5**(1): p. 477-483.
16. Deng, J., Li, M., and Wang, Y., *Biomass-derived carbon: synthesis and applications in energy storage and conversion*. Green Chemistry, 2016. **18**(18): p. 4824-4854.
17. Han, J., Zhang, L., Zhao, B., Qin, L., Wang, Y., and Xing, F., *The N-doped activated carbon derived from sugarcane bagasse for CO₂ adsorption*. Industrial Crops and Products, 2019. **128**: p. 290-297.
18. Lin, Y., Chen, Z., Yu, C., and Zhong, W., *Heteroatom-Doped Sheet-Like and Hierarchical Porous Carbon Based on Natural Biomass Small Molecule Peach Gum for High-Performance Supercapacitors*. ACS Sustainable Chemistry & Engineering, 2019. **7**(3): p. 3389-3403.
19. Kwiatkowski, J.F., *Activated Carbon : Classifications, Properties and Applications*. Nova Science Publishers, Inc. 2012.
20. Linares-Solano, A., Lillo-Ródenasm, M.A., Marco-Lozar, J.P., Kunowsky, M., and Romero-Anaya, A.J., *NaOH and KOH form Preparing Activating Carbons Used in Energy and Environmental Applications*. 2012.
21. Ahmed, M.J. and Theydan, S.K., *Optimization of microwave preparation conditions for activated carbon from Albizia lebbeck seed pods for methylene blue dye adsorption*. Journal of Analytical and Applied Pyrolysis, 2014. **105**: p. 199-208.
22. Khan, T.A., Saud, A.S., Jamari, S.S., Rahim, M.H.A., Park, J.-W., and Kim, H.-J., *Hydrothermal carbonization of lignocellulosic biomass for carbon rich material preparation: A review*. Biomass and Bioenergy, 2019. **130**.
23. Susanti, R.F., Kevin, G., Erico, M., Andreas, A., Kristianto, H., and Handoko, T., *Delignification, Carbonization Temperature and Carbonization Time Effects on the Hydrothermal Conversion of Salacca Peel*. J Nanosci Nanotechnol, 2018. **18**(10): p. 7263-7268.
24. Yan, S., Lin, J., Liu, P., Zhao, Z., Lian, J., Chang, W., Yao, L., Liu, Y., Lin, H., and Han, S., *Preparation of nitrogen-doped porous carbons for high-performance*

- supercapacitor using biomass of waste lotus stems.* RSC Advances, 2018. **8**(13): p. 6806-6813.
25. Chen, J., Yang, J., Hu, G., Hu, X., Li, Z., Shen, S., Radosz, M., and Fan, M., *Enhanced CO₂ Capture Capacity of Nitrogen-Doped Biomass-Derived Porous Carbons.* ACS Sustainable Chemistry & Engineering, 2016. **4**(3): p. 1439-1445.
26. Ahmed, S., Rafat, M., and Ahmed, A., *Nitrogen doped activated carbon derived from orange peel for supercapacitor application.* Advances in Natural Sciences: Nanoscience and Nanotechnology, 2018. **9**(3).
27. Bansal, R.C. and Goyal, M., *Activated Carbon Adsorption, Chromatographia.* 2005.
28. Ruthven, D.M., *Principles of Adsorption and Adsorption Processes.* 1984. p. 454.
29. Chandra, R.P., Bura, R., Mabee, W.E., Berlin, A., Pan, X., and Saddler, J.N., *Substrate pretreatment: the key to effective enzymatic hydrolysis of lignocellulosics?* Adv Biochem Eng Biotechnol, 2007. **108**: p. 67-93.
30. Agbor, V.B., Cicek, N., Sparling, R., Berlin, A., and Levin, D.B., *Biomass pretreatment: fundamentals toward application.* Biotechnol Adv, 2011. **29**(6): p. 675-85.
31. Hendriks, A.T. and Zeeman, G., *Pretreatments to enhance the digestibility of lignocellulosic biomass.* Bioresour Technol, 2009. **100**(1): p. 10-8.
32. Kumar, P., Barrett, D.M., Delwiche, M.J., and Stroeve, P., *Methods for Pretreatment of Lignocellulosic Biomass for Efficient Hydrolysis and Biofuel Production.* Industrial & Engineering Chemistry Research, 2009. **48**(8): p. 3713-3729.
33. Wang, T., Zhai, Y., Zhu, Y., Li, C., and Zeng, G., *A review of the hydrothermal carbonization of biomass waste for hydrochar formation: Process conditions, fundamentals, and physicochemical properties.* Renewable and Sustainable Energy Reviews, 2018. **90**: p. 223-247.
34. Sevilla, M. and Fuertes, A.B., *The production of carbon materials by hydrothermal carbonization of cellulose.* Carbon, 2009. **47**(9): p. 2281-2289.
35. Jagtoyen, M. and Frank, D., *Activated carbons from Yellow Poplar and White Oak by H₃PO₄ Activation.* 1998. **36**(7-8): p.1085-1097.
36. Palmqvist, E. and Hahn-Hägerdal, B., *Fermentation of lignocellulosic hydrolysates. II: inhibitors and mechanisms of inhibition.* Bioresource Technology, 2000. **74**(1): p. 25-33.

37. Kang, S., Li, X., Fan, J., and Chang, J., *Characterization of Hydrochars Produced by Hydrothermal Carbonization of Lignin, Cellulose, d-Xylose, and Wood Meal*. Industrial & Engineering Chemistry Research, 2012. **51**(26): p. 9023-9031.
38. Sun, Y. and Cheng, J., *Hydrolysis of Lignocellulosic Materials for Ethanol Production : A Review*. 2002.
39. Manurung, R., Syahputra, A., Alhamdi, M.A., Satria, W., Barus, E.M., Hasibuan, R., and Siswarni, M.Z., *Delignification and Hydrolysis Lignocellulosic of Bagasse in Choline Chloride System*. IOP Conference Series: Earth and Environmental Science, 2018. **122**.
40. Zheng, Y., Zhao, J., Xu, F., and Li, Y., *Pretreatment of lignocellulosic biomass for enhanced biogas production*. Progress in Energy and Combustion Science, 2014. **42**: p. 35-53.
41. Chen, H., Liu, J., Chang, X., Chen, D., Xue, Y., Liu, P., Lin, H., and Han, S., *A review on the pretreatment of lignocellulose for high-value chemicals*. Fuel Processing Technology, 2017. **160**: p. 196-206.
42. Marsh, H., and Fransisco, R.-R., *Activated Carbon*. 2006.
43. Li, Z., Pan, Z., and Zhang, R., *Overview of biomass pretreatment for cellulosic ethanol production*. International Journal of Agricultural and Biological Engineering. 2009. **2**(3): p. 51-68.
44. Park, S., Baker, J.O., Himmel, M.E., Parilla, P.A., and Johnson, D.K., *Cellulosa Crystallinity Index : Measurement Techniques and Their Impact on Interpreting Cellulase Performance*. Biotechnology for Biofuels. 2010. **3**: p. 1-10.
45. Zhu, J.Y., Wang, G.S., Pan, X.J., and Gleisner, R., *Specific surface to evaluate the efficiencies of milling and pretreatment of wood for enzymatic saccharification*. Chemical Engineering Science, 2009. **64**(3): p. 474-485.
46. Kim, J.S., Lee, Y.Y., and Kim, T.H., *A review on alkaline pretreatment technology for bioconversion of lignocellulosic biomass*. Bioresour Technol, 2016. **199**: p. 42-48.
47. Funke, A. and Ziegler, F., *Hydrothermal carbonization of biomass: A summary and discussion of chemical mechanisms for process engineering*. Biofuels, Bioproducts and Biorefining, 2010. **4**(2): p. 160-177.

48. Sevilla, M. and Mokaya, R., *Energy storage applications of activated carbons: supercapacitors and hydrogen storage*. Energy Environ. Sci., 2014. **7**(4): p. 1250-1280.
49. Jain, A., Balasubramanian, R., and Srinivasan, R., *Hydrothermal conversion of biomass waste to activated carbon with high porosity: A review*. Chemical Engineering Journal, 2016. **283**: p. 789-805.
50. Libra, J.A., Ro, K.S., Kammann, C., Funke, A., Nicole, D.B., Neubauer, Y., Titirici, M.-M., Fuhner, C., Bens, O., Kern, J., and Emmerich, K.-H., *Hydrothermal Carbonization of Biomass Residuals : a comparative Review of The Chemistry, Processes and Applications of Wet and Dry Pyrolysis*. Biofuels. 2011. **2**(1): p. 71-106.
51. Junting, Z., Guangming, L., Wenzhi, H., Juwen, H., and Haochen, Z., *Hydrothermal carbonization (HTC) for recovery of organic fractions in municipal solid waste (OFMSW)*. 2017.
52. Moller, M., Nilges, P., Harnisch, F., and Schroder, U., *Subcritical water as reaction environment: fundamentals of hydrothermal biomass transformation*. ChemSusChem, 2011. **4**(5): p. 566-79.
53. Coronella, C.J., Lynam, J.G., Reza, M.T., and Uddin, M.H., *Hydrothermal Carbonization of Lignocellulosic Biomass*, in *Application of Hydrothermal Reactions to Biomass Conversion*. 2014. p. 275-311.
54. Fang, J., Zhan, L., Ok, Y.S., and Gao, B., *Minireview of potential applications of hydrochar derived from hydrothermal carbonization of biomass*. Journal of Industrial and Engineering Chemistry, 2018. **57**: p. 15-21.
55. Nizamuddin, S., Baloch, H.A., Griffin, G.J., Mubarak, N.M., Bhutto, A.W., Abro, R., Mazari, S.A., and Ali, B.S., *An overview of effect of process parameters on hydrothermal carbonization of biomass*. Renewable and Sustainable Energy Reviews, 2017. **73**: p. 1289-1299.
56. Zhang, T., Walawender, W., Fan, L., Fan, M., Daugaard, D., and Brown, R., *Preparation of activated carbon from forest and agricultural residues through CO activation*. Chemical Engineering Journal, 2004. **105**(1-2): p. 53-59.
57. Zhang, Y.-J., Xing, Z.-J., Duan, Z.-K., Meng, L., and Wang, Y., *Effects of steam activation on the pore structure and surface chemistry of activated carbon derived from bamboo waste*. Applied Surface Science, 2014. **315**: p. 279-286.

58. Yorgun, S., Vural, N., and Demiral, H., *Preparation of high-surface area activated carbons from Paulownia wood by ZnCl₂ activation*. Microporous and Mesoporous Materials, 2009. **122**(1-3): p. 189-194.
59. Dabrowski, A., *Adsorption and its Application in Industry and Environmental Protection*. 1998. **120**(A).
60. Rodriguez-Reinozo, F. and Molina-Sabio, M., *Activated Carbon from Lignocellulosic Materials by Chemical and/or Physical Activation : an Overview*. Carbon. 1992. **30**(7): p. 1111-1118.
61. Hayashi, J., Kazehaya, A., Muroyama, K., and Watkinson, A.P., *Preparation of Activated Carbon from Lignin by Chemical Activation*. Carbon. 2000. **38**:p. 1873-1878.
62. Liu, B., Gu, J., and Zhou, J., *High surface area rice husk-based activated carbon prepared by chemical activation with ZnCl₂-CuCl₂composite activator*. Environmental Progress & Sustainable Energy, 2016. **35**(1): p. 133-140.
63. Huang, Y., Liu, Z., and Zhao, G., *Reaction process for ZnCl₂activation of phenol liquefied wood fibers*. RSC Adv., 2016. **6**(82): p. 78909-78917.
64. Heidarinejad, Z., Dehghani, M.H., Heidari, M., Javedan, G., Ali, I., and Sillanpää, M., *Methods for preparation and activation of activated carbon: a review*. Environmental Chemistry Letters, 2020. **18**(2): p. 393-415.
65. Mi, T., Chen, L., Xin, S.-Z., and Yu, X.-M., *Activated Carbon from the Chinese Herbal Medicine Waste by H₃PO₄Activation*. Journal of Nanomaterials, 2015. **2015**: p. 1-9.
66. Wang, J. and Kaskel, S., *KOH activation of carbon-based materials for energy storage*. Journal of Materials Chemistry, 2012. **22**(45).
67. Sevilla, M., Yu, L., Ania, C.O., and Titirici, M.-M., *Supercapacitive Behavior of Two Glucose-Derived Microporous Carbons: Direct Pyrolysis versus Hydrothermal Carbonization*. ChemElectroChem, 2014. **1**(12): p. 2138-2145.
68. Vangari, M., Pryor, T., and Jiang, L., *Supercapacitors: Review of Materials and Fabrication Methods*. Journal of Energy Engineering, 2013. **139**(2): p. 72-79.
69. González-García, P., *Activated carbon from lignocellulosics precursors: A review of the synthesis methods, characterization techniques and applications*. Renewable and Sustainable Energy Reviews, 2018. **82**: p. 1393-1414.

70. Jagadale, A., Zhou, X., Xiong, R., Dubal, D.P., Xu, J., and Yang, S., *Lithium ion capacitors (LICs): Development of the materials.* Energy Storage Materials, 2019. **19**: p. 314-329.
71. Xia, K., Gao, Q., Jiang, J., and Hu, J., *Hierarchical porous carbons with controlled micropores and mesopores for supercapacitor electrode materials.* Carbon, 2008. **46**(13): p. 1718-1726.
72. Barcellona, S., Ciccarelli, F., Iannuzzi, D., and Piegari, L., *Overview of Lithium-ion Capacitor Applications Based on Experimental Performances.* Electric Power Components and Systems, 2016. **44**(11): p. 1248-1260.
73. Davies, A. and Yu, A., *Material advancements in supercapacitors: From activated carbon to carbon nanotube and graphene.* The Canadian Journal of Chemical Engineering, 2011. **89**(6): p. 1342-1357.
74. Stoller, M.D., Park, S., Zhu, Y., An, J., and Ruoff, R.S., *Graphene-Based Ultracapacitors.* 2008. pp. 6-10.
75. Mandic, G., Nasiri, A., Ghotbi, E., and Muljadi, E., *Lithium-Ion Capacitor Energy Storage Integrated with Variable Speed Wind Turbines for Power Smoothing.* IEEE Journal of Emerging and Selected Topics in Power Electronics. 2013. **1**(4): p. 287-295.
76. Sato, A., *TAIYO YUDEN Lithium Ion Capacitors: An Effective EDLC Replacement.*
77. Li, G., Yang, Z., Yin, Z., Guo, H., Wang, Z., Yan, G., Liu, Y., Li, L., and Wang, J., *Non-aqueous dual-carbon lithium-ion capacitors: a review.* Journal of Materials Chemistry A, 2019. **7**(26): p. 15541-15563.
78. Scherdel, C., Reichenauer, G., and Wiener, M., *Relationship between pore volumes and surface areas derived from the evaluation of N₂-sorption data by DR-, BET- and t-plot.* Microporous and Mesoporous Materials, 2010. **132**(3): p. 572-575.
79. Lowell, S., Shields, J.E., Thomas, M.A., and Thommes, M., *Characterizations of Porous Solids and Powders : Surface Area, Pore Size and Density.* 2004.
80. Gelb, L.D., and Gubbins, K.E., *Characterization of Porous Glasses : Simualtion Models, Adsorption Isotherms, and the Brunauer-Emmett-Teller Analysis Method.* Langmuir. 1998. **14**(8): p. 2097-2111.
81. Naderi, M., *Surface Area, in Progress in Filtration and Separation.* 2015. p. 585-608.

82. Khan, M.S.I., Oh, S.W., and Kim, Y.J., *Power of Scanning Electron Microscopy and Energy Dispersive X-Ray Analysis in Rapid Microbial Detection and Identification at the Single Cell Level*. Sci Rep, 2020. **10**(1): p. 2368.
83. Gazulla, M.F., Rodrigo, M., Blasco, E., and Orduña, M., *Nitrogen determination by SEM-EDS and elemental analysis*. X-Ray Spectrometry, 2013. **42**(5): p. 394-401.
84. Akhtar, K., Khan, A., Khan, S.B., and Asiri, A.M., *Scanning Electron Microscopy: Principle and Applications in Nanomaterials Characterization*, in *Handbook of Materials Characterization*. 2018. p. 113-145.
85. Ravichandran, P., Sugumaran, P., Seshadri, S., and Basta, A.H., *Optimizing the route for production of activated carbon from Casuarina equisetifolia fruit waste*. R Soc Open Sci, 2018. **5**(7): p. 171578.
86. Zhu, F.Y., Wang, Q.Q., Zhang, X.S., Hu, W., Zhao, X., and Zheng, H.X., *3D nanostructure reconstruction based on the SEM imaging principle, and applications*. Nanotechnology, 2014. **25**(18): p. 185705.
87. Ingemarsson, L. and Halvarsson, M., *SEM-EDX Analysis of Boron*. 2011. p. 1-15.
88. Vlachos, N., Skopelitis, Y., Psaroudaki, M., Konstantinidou, V., Chatzilazarou, A., and Tegou, E., *Applications of Fourier transform-infrared spectroscopy to edible oils*. Anal Chim Acta, 2006. **573-574**: p. 459-65.
89. Petit, S. and Madejova, J., *Fourier Transform Infrared Spectroscopy*, in *Handbook of Clay Science*. 2013. p. 213-231.
90. Nicolet, *Introduction Fourier Transform Infrared Spectrometry*. Thermo Nicolet Corporation. 2001. p. 1-8.
91. Das, R., Ali, M.E., and Hamid, S.B.A., *Current applications of x-ray powder diffraction - A review*. Review on Advanced Materials Science. 2014. **38**(2): p. 95-109.
92. Bunaciu, A.A., Udristioiu, E.G., and Aboul-Enein, H.Y., *X-ray diffraction: instrumentation and applications*. Crit Rev Anal Chem, 2015. **45**(4): p. 289-99.
93. Fan, F., Yang, Z., Li, H., Shi, Z., and Kan, H., *Preparation and properties of hydrochars from macadamia nut shell via hydrothermal carbonization*. R Soc Open Sci, 2018. **5**(10): p. 181126.
94. Rodriguez Correa, C., Hehr, T., Voglhuber-Slavinsky, A., Rauscher, Y., and Kruse, A., *Pyrolysis vs. hydrothermal carbonization: Understanding the effect of biomass*

- structural components and inorganic compounds on the char properties.* Journal of Analytical and Applied Pyrolysis, 2019. **140**: p. 137-147.
95. Alhnidi, M.J., Korner, P., Wust, D., Pfersich, J., and Kruse, A., *Nitrogen-Containing Hydrochar: The Influence of Nitrogen-Containing Compounds on the Hydrochar Formation.* ChemistryOpen, 2020. **9**(8): p. 864-873.
 96. Kruse, A., Koch, F., Stelzl, K., Wüst, D., and Zeller, M., *Fate of Nitrogen during Hydrothermal Carbonization.* Energy & Fuels, 2016. **30**(10): p. 8037-8042.
 97. Straten, J.W., Schleker, P., Krasowska, M., Veroutis, E., Granwehr, J., Auer, A.A., Hetaba, W., Becker, S., Schlogl, R., and Heumann, S., *Nitrogen-Functionalized Hydrothermal Carbon Materials by Using Urotropine as the Nitrogen Precursor.* Chemistry, 2018. **24**(47): p. 12298-12317.
 98. Kristianto, H., *REVIEW: SINTESIS KARBON AKTIF DENGAN MENGGUNAKAN AKTIVASI KIMIA ZnCL2.* Jurnal Integrasi Proses, 2017. **6**(3).
 99. Xiao, K., Liu, H., Li, Y., Yang, G., Wang, Y., and Yao, H., *Excellent performance of porous carbon from urea-assisted hydrochar of orange peel for toluene and iodine adsorption.* Chemical Engineering Journal, 2020. **382**.
 100. Tian, K., Wu, Z., Xie, F., Hu, W., and Li, L., *Nitrogen-Doped Porous Carbons Derived from Triarylisocyanurate-Cored Polymers with High CO₂ Adsorption Properties.* Energy & Fuels, 2017. **31**(11): p. 12477-12486.
 101. Bai, K., Hao, J., Yang, Y., and Qian, A., *The effect of hydrothermal temperature on the properties of SBA-15 materials.* Heliyon, 2020. **6**(8): p. e04436.
 102. Rodríguez Correa, C., Stollovsky, M., Hehr, T., Rauscher, Y., Rolli, B., and Kruse, A., *Influence of the Carbonization Process on Activated Carbon Properties from Lignin and Lignin-Rich Biomasses.* ACS Sustainable Chemistry & Engineering, 2017. **5**(9): p. 8222-8233.
 103. Bruice, P.Y., *Organic Chemistry EIGHTH EDITION.* Pearson Education. 2017.
 104. Silverstein, R.M., Webster, F.X.W., and Kiemle, D., *Spectrometric Identification of Organic Compounds.* 2005.
 105. Kurniawan, R.Y., Kurniawan, I.D.O, Atmaja, L., and Widiastuti, N., *Synthesis N-Doped Activated Carbon from Sugarcane Bagasse for CO₂ Adsorption.* IPTEK The Journal for Technology and Science. 2019. **30**(3): p. 80.

106. Nunes, C., Mahendrasingam, A., and Suryanarayanan, R., *Quantification of crystallinity in substantially amorphous materials by synchrotron X-ray powder diffractometry*. Pharm Res, 2005. **22**(11): p. 1942-53.
107. Susanti, R.F., Alvin, S., and Kim, J-H., *Toward high-performance hard carbon as an anode for sodium-ion batteries: Demineralization of biomass as a critical step*. Journal of Industrial and Engineering Chemistry. 2020. **91**(8): p. 317-329.